

## MATERIALS FOR ELECTRICAL ENERGY STORAGE

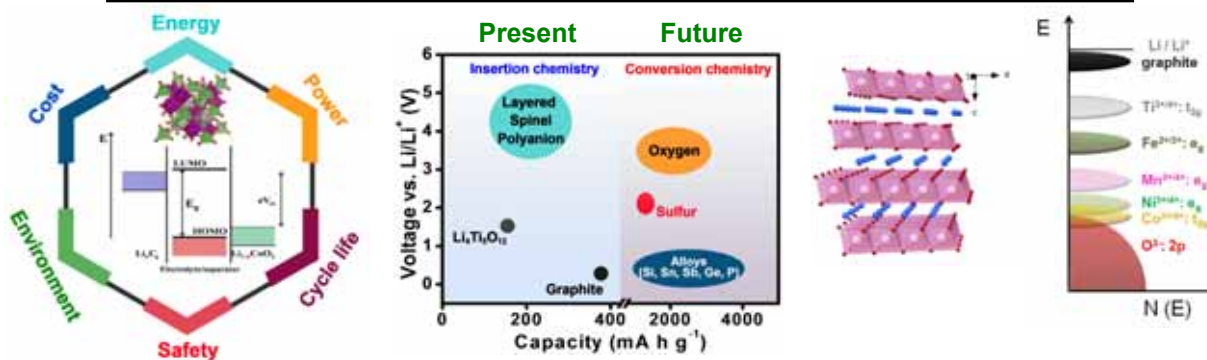
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## USE-INSPIRED BASIC SCIENCE RESEARCH WITH ENGINEERING TARGETS



- Renewable energy is the solution to address the world's energy and environmental challenges
- Storage is the bottleneck to utilize efficiently and economically the intermittent renewable energy
- We are engaged in the development of energy-dense, long-life, safe batteries at an affordable cost
  - cobalt-free lithium-ion and sodium-ion batteries; lithium-sulfur and sodium-sulfur batteries
  - all-solid-state lithium and sodium batteries; solid-electrolyte redox flow batteries

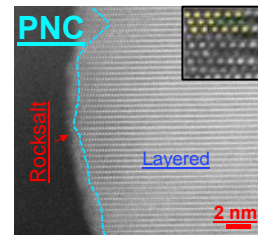
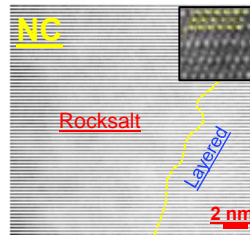
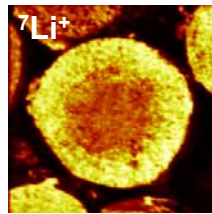
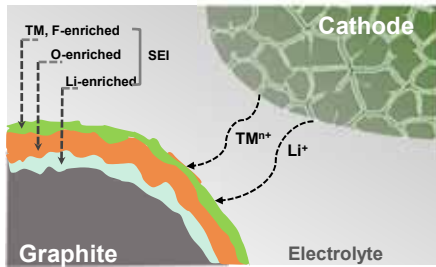
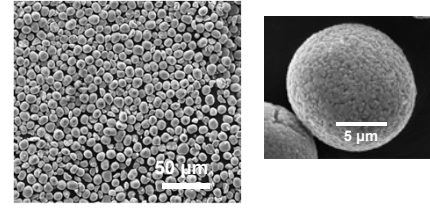
A. Manthiram, *ACS Central Science* **3**, 1063 (2017); A. Manthiram, *Nature Communications* **11**, 1550 (2020)

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## SYNTHESIS INNOVATION AND ADVANCED CHARACTERIZATION

- Controlled synthesis with practically relevant characteristics
  - scale-up: 10 kg per batch, incorporation of hard to dope ions
- Pouch full cell fabrication with lithium-ion, sodium-ion, Li-S
- In-situ and ex-situ XRD, SEM, TEM, XPS, TOF-SIMS, etc.
  - before and after extended full pouch cell cycling
- In-depth electrochemical characterization of pouch full cells



W. Li, X. Liu, H. Celio, P. Smith, A. Dolocan, M. Chi, and A. Manthiram, *Advanced Energy Materials* **8**, 1703154:1-11 (2018)  
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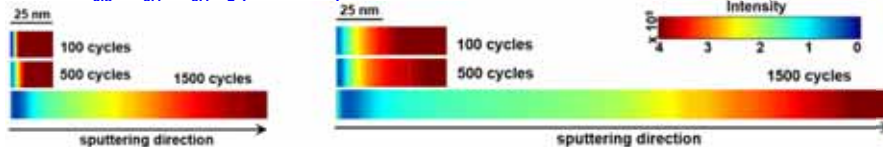
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## GRAPHITE-ELECTROLYTE INTERPHASE EVOLUTION WITH TOF-SIMS

### Depth Profile of C<sub>5</sub><sup>-</sup> fragment (representing bulk graphite)

Graphite paired with LiNi<sub>0.8</sub>Mn<sub>0.1</sub>Co<sub>0.1</sub>O<sub>2</sub> (NMC 811)

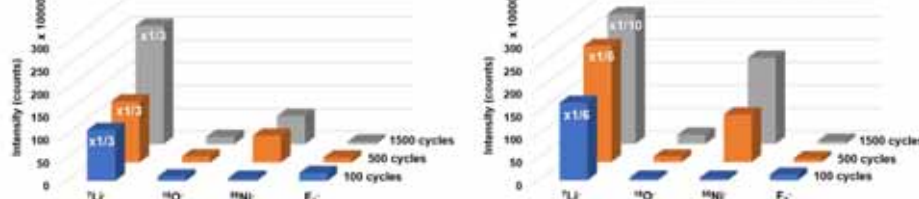
Graphite paired with LiNi<sub>0.94</sub>Co<sub>0.06</sub>O<sub>2</sub> (NC 9406)



- AEI thickness can be estimated by probing C<sub>5</sub><sup>-</sup> (representing graphite) with sputtering
- AEI on graphite paired with NC 9406 is much thicker than that paired with NMC 811

Graphite paired with LiNi<sub>0.8</sub>Mn<sub>0.1</sub>Co<sub>0.1</sub>O<sub>2</sub> (NMC 811)

Graphite paired with LiNi<sub>0.94</sub>Co<sub>0.06</sub>O<sub>2</sub> (NC 9406)

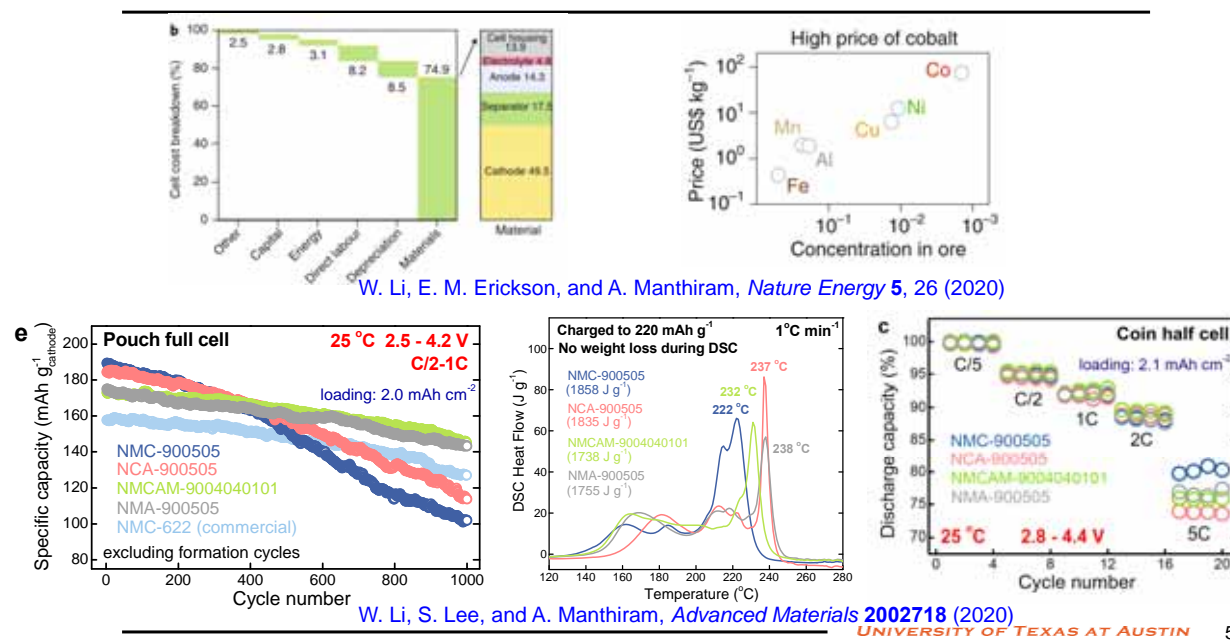


- Continuous accumulation of transition-metal ions & trapping of active Li on graphite surface
- NC 9406 exhibits more severe transition-metal dissolution & leads to Li trapping on graphite

J. Li and A. Manthiram, *Advanced Energy Materials* **9**, 1902731 (2019)

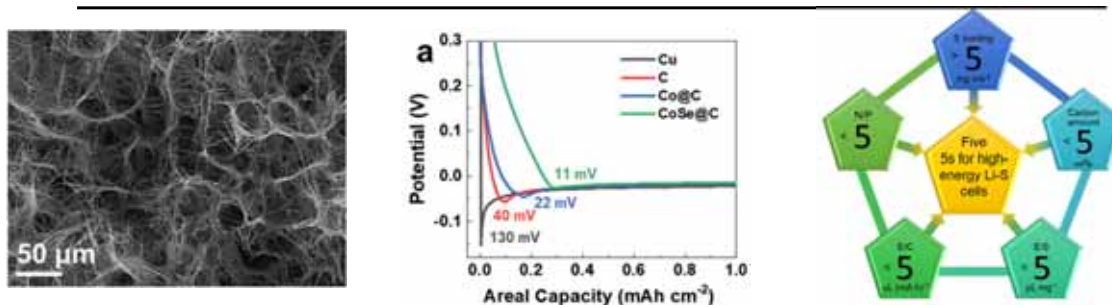
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## COBALT-FREE CATHODE: HIGH NICKEL $\text{LiNi}_{0.9}\text{Mn}_{0.05}\text{Al}_{0.05}\text{O}_2$ (NMA-900505)



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## LITHIUM-SULFUR BATTERIES WITH STABILIZED LITHIUM METAL



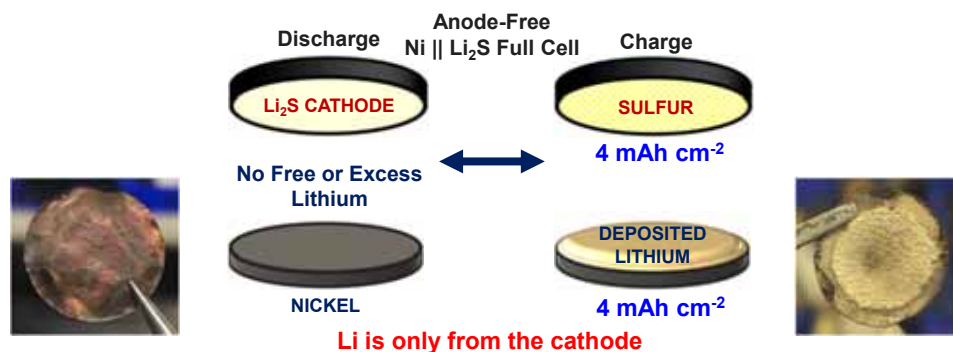
- Sulfur is abundant and inexpensive, but met with challenges for lithium deposition and cycle life
- Can we develop a better understanding of the dynamics of lithium deposition?
- Can we engineer a uniform SEI layer *in situ* or a 3D framework with favorable properties?
- What role do the crossed over species play on the dynamics of lithium deposition?

J. He and Manthiram, *Advanced Energy Materials* 2002654 (2020)  
 A. Bhargav, J. He, A. Gupta, and A. Manthiram, *Joule* 4, 285-291 (2020)  
 A. Manthiram, Y.-Z. Fu, S.-H. Chung, C. Zu, & Y.-S. Su, *Chemical Reviews* 114, 11751 (2014)

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## ANODE-FREE FULL CELL TO UNDERSTAND THE DYNAMICS



- Serves as an excellent template for evaluating the dynamics of lithium deposition
- Maximizes gravimetric and volumetric energy densities
- Eliminates self-discharge in assembled cells as they are in discharged state
- Obviates the technical challenges associated with thin lithium foils

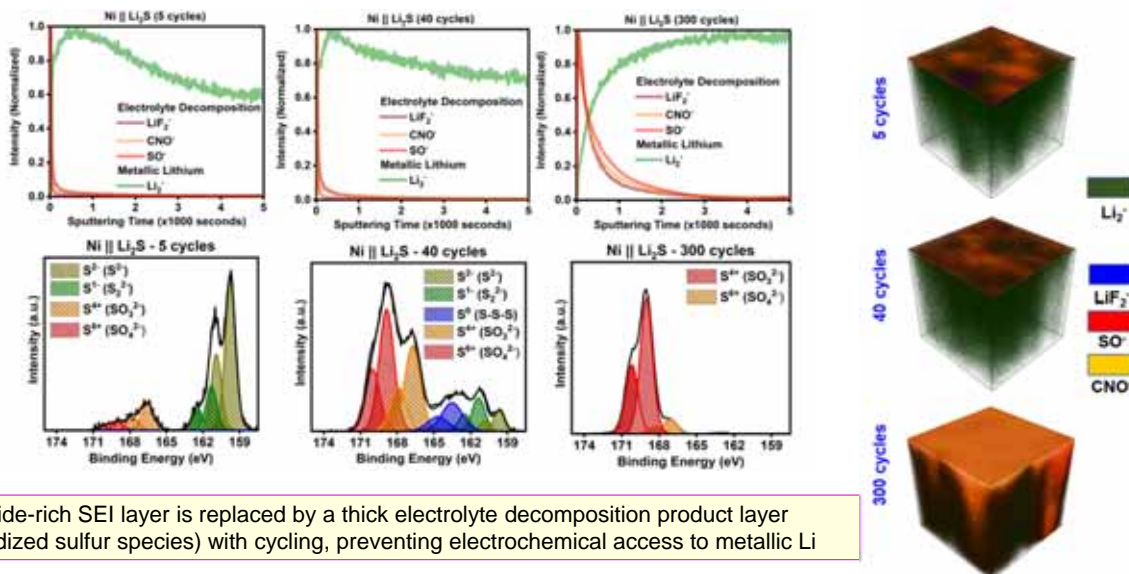
S. Nanda, A. Gupta, and A. Manthiram. *Advanced Energy Materials* 8, 1801556 (2018)

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## WHAT CAUSES LITHIUM-METAL DEGRADATION?



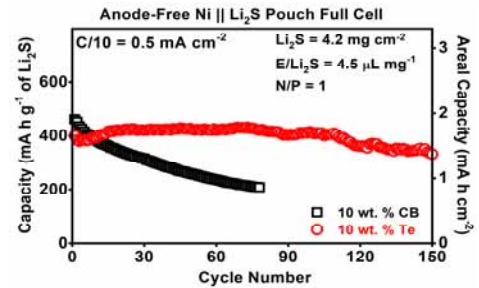
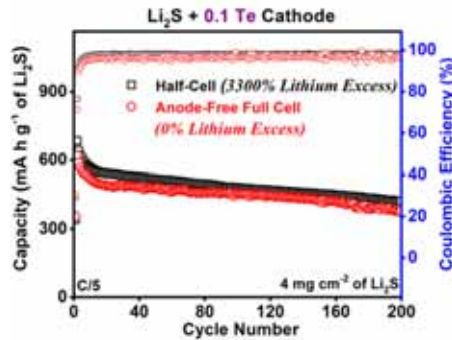
S. Nanda and A. Manthiram, *Energy & Environmental Science*, 13, 2501 (2020)

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## HOW DO WE SUPPRESS LITHIUM-METAL DEGRADATION?



Tellurium addition to sulfur cathode

- forms polytellurosulfides ( $\text{Li}_2\text{Te}_x\text{S}_y$ ) and deposits  $\text{Li}_2\text{TeS}_3$  layer on lithium-metal anode
- suppresses electrolyte decomposition and helps retain good cycle life
- allows operation of practically relevant pouch cells with lean electrolyte

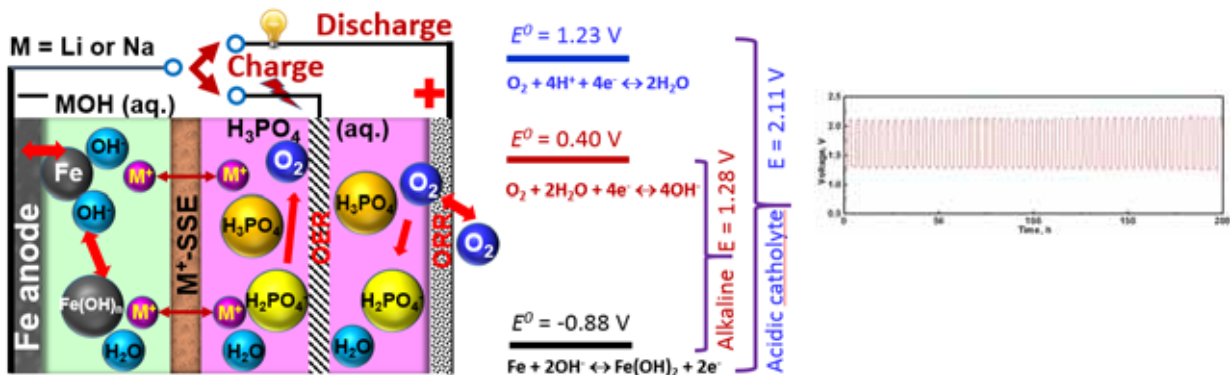
S. Nanda, A. Bhargav, and A. Manthiram, *Joule* 4,1121 (2020)

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## MEDIATOR-ION SOLID-ELECTROLYTE FLOW BATTERIES



- Solid electrolyte eliminates chemical crossover, suppresses dendrite (e.g., zincate migration, carbonation by  $\text{CO}_2$ )
- Allows different media at the anode and cathode (aqueous vs. nonaqueous, acidic vs. basic, solid vs. liquid, etc.)

L. Li and A. Manthiram, *Advanced Energy Materials* 5, 1502054 (2015)

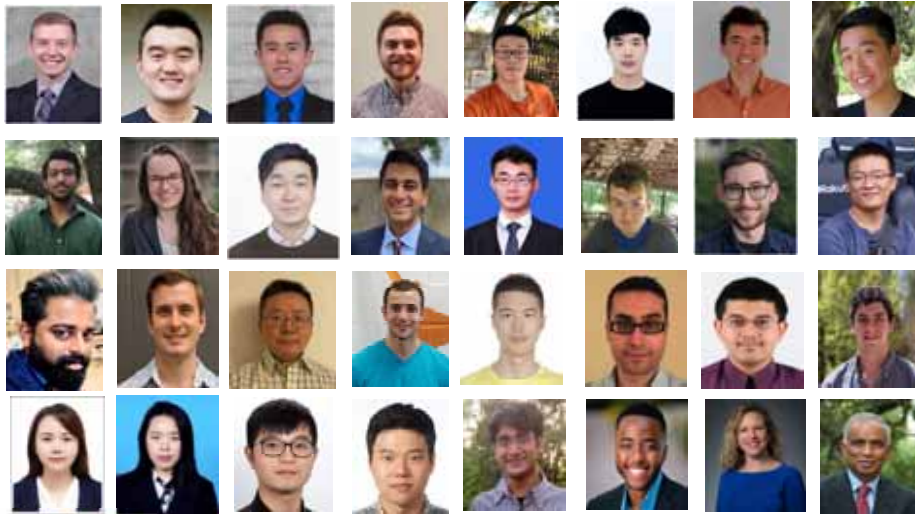
X. Yu and A. Manthiram, *ACS Energy Letters* 2, 1050 (2017)

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Large group  
Interdisciplinary  
Resources  
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Training  
Mentoring  
Motivation  
Passion



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